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The effect of CO₂-philic thickeners on gravity drainage mechanism in gas invaded zone

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The effect of CO₂-philic thickeners on gravity drainage mechanism in gas invaded zone

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Abstract

The rate of mass transfer between the fractures and matrix in gas invaded zone can significantly influence on the oil recovery during the forced gravity drainage process. However, in this study, a new approach was suggested to improve the gravity drainage process in gas invaded zone. Poly(fluoroacrylate) (PFA), as a CO₂-philic thickener, was injected into the gas invaded zone to illustrate the impact of interfacial mechanisms such as gas diffusion coefficient and interfacial tension (IFT) on oil recovery. Also, the cloud point pressures were measured to ensure that the PFA did not come out of the solution due to a phase change during IFT, gas diffusion coefficient, and gravity drainage experiments. Results showed that the CO₂-PFA thickener (20000 ppm) could decrease the IFT from 56 to 24 dyne/cm compared to the pure CO₂ scenario, improving the gravity drainage mechanism in the gas invaded zone. In addition, the CO₂ diffusion coefficients were increased approximately more than two times during CO₂-PFA injection in comparison with pure CO₂ injection in both porous media and bulk oil phase scenarios at reservoir

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27 conditions. Also, an incremental oil recovery of 16 percent was achieved during PFA/CO₂
28 compared to pure CO₂ injection in the gas invaded zone. Therefore, gas gravity drainage is the
29 most important mechanism once gas thickener or CO₂ enters the fractures in the gas invaded
30 zone.

31 **Keywords:** Gas invaded zone, Gas thickeners, CO₂ diffusion coefficients, Gravity drainage,
32 Matrix-fracture system, Cloud point pressure

33 **Introduction**

34 CO₂ injection in naturally fracture reservoirs is the process of storing and capturing atmospheric
35 carbon dioxide combined with enhanced oil recovery. It is one method of decreasing the amount
36 of CO₂ in the atmosphere with the goal of reducing global climate change ([Hassanpouryouzband
37 et al., 2021 and 2018](#)). Naturally fractured carbonate reservoirs (NFRs) hold a significant portion
38 of the global oil reserves. Gravity is the only conceivable economic driving mechanism in widely
39 fractured reservoirs because extremely conductive fractures create shortcuts for the injected
40 fluids. In the production from fractured carbonate reservoirs, three zones can be recognised: (1) a
41 gas invaded zone with oil-filled matrix and gas-filled fractures, (2) an oil rim with oil-filled
42 matrix and fractures, and (3) a water invaded zone with water-filled fractures and oil-filled
43 matrix ([Li et al., 2018](#); [Kharrat et al., 2021](#); [Farrokhrouz et al., 2022](#)). The oil flow in the gas
44 invaded zone will be largely through the matrix due to the gravity drainage mechanism. When
45 gas from gas-saturated fractures moves oil in the matrix, gas gravity drainage occurs. The gas
46 gravity drainage is derived from the difference in density of the oil and gas phases. By injecting
47 gas or gas thickeners into the fracture-matrix system, these contact with insitu fluid through
48 diffusion. This makes the oil to swell, and consequently lessens the oil viscosity, absorbs the

49 light component, and drops the IFT. Therefore, the gas diffusion coefficient and IFT can affect
50 the gravity drainage mechanism in the gas invaded zone. Therefore, it is significantly important
51 to fully understand these mechanisms under actual reservoir conditions (Guo et al., 2022;
52 Aghabarari et al., 2022). The diffusion coefficient is a key parameter in controlling the mixing
53 rate of the injected gas and insitu fluids. Gas dissolution will change the insitu fluid properties
54 significantly including the reduction in oil viscosity and oil swelling which consequently result
55 in the improvement of oil mobility and enhancing the oil recovery (Gao et al., 2019). In Berea
56 cores saturated with n-hexadecane, Li et al. (2006 and 2009) estimated the effective CO₂
57 diffusion coefficient. They came to the conclusion that the readings were slightly affected by
58 pressure changes ranging from 2.3 to 6.3 MPa. Li et al. (2016) investigated the impacts of oil
59 saturation and tortuosity on CO₂ diffusivity in porous media with poor permeability. The
60 diffusion coefficients were found to be highly influenced by the oil saturation and permeability
61 of the porous medium. Gao et al. (2019) used tortuosity to link the CO₂ mass transfer coefficient
62 to the permeability of porous media. They discovered that the high tortuosity can slow down the
63 CO₂ diffusivity by restricting gas solubility. The prior measurements of diffusion coefficients, on
64 the other hand, still need to be extended by using gas thickeners. In addition, several field
65 performance analyses (Al-Shibli et al., 2022; Zobeidi et al., 2021; King et al., 1970; Carlson,
66 1988) and laboratory experiments (Saidi et al., 1993; Clemens et al., 2001; Sajjadian et al., 1999,
67 Zobeidi et al., 2018) showed that the oil recovery factor could be significantly enhanced if
68 gravity drainage process is the leading production mechanism. Carlson (1988) also came to the
69 conclusion that high oil recovery after a gas drive operation is attributable to gravity forces
70 reflected by considerable differences in density between the oil and gas. However, the obtained
71 conclusion ignores the process's numerous side consequences. Laboratory studies of gravity

72 drainage in fractured rock were conducted by Sajjadian et al. (1999). They reported that capillary
73 continuity and re-infiltration could improve oil recovery factor. Clemens et al. (2001)
74 demonstrated that when there are no fractures in the horizontal plane and the oil must flow
75 sideways into the fracture system, the drainage rate is further lowered, resulting in lower
76 drainage rates. Ameri et al. (2013) investigated the effect of miscibility on the gas-
77 oil gravity drainage in naturally fractured reservoirs. The findings show that injecting a non-
78 equilibrium gas with a larger scale of solubility into the oil phase causes a zone of reduced oil
79 viscosity, leading to enhanced gravity-mediated recovery. Zobeidi et al. (2018) investigated the
80 impact of gravity drainage on the block to block interactions in NFRs. Their findings showed
81 that oil penetration into lower blocks occurs rapidly. Hasanzadeh et al. (2021) investigated the
82 forced and free-fall gravity drainage mechanism in a fractured physical model. Results reveal
83 that the forced gravity drainage performs better under controlled process conditions than the
84 free-fall gravity drainage. Karimaie et al. (2010) investigated gas-oil gravity drainage process in
85 fractured carbonate rock with low IFT and gas injection. They showed that even after water
86 injection, low IFT gravity drainage may recover a large amount of oil in NFRs. Ameri et al.
87 (2015) evaluated the rate of mass-transfer between the fractures and the matrix while a gas
88 solvent is injected into a fracture system. Their findings revealed that matrix wettability has no
89 effect on solvent injection performance, and that the remaining oil in the matrix may be
90 recovered using an increased gravity drainage procedure. Kahrobaei et al. (2012) illustrated that
91 transfer rates of solvent between fracture and matrix is a function of the rock permeability, oil
92 viscosity and the density of both oil and solvent. Consequently, based on the previous works, the
93 data of CO₂ diffusion coefficients and gravity drainage mechanism in hydrocarbon fluids and
94 crude oil is still undersupplied at the actual reservoir conditions. Also, a large portion of the

95 recorded data are within a limited temperature range ($<60\text{ }^{\circ}\text{C}$) and pressures ($<2500\text{ psi}$), and the
96 effect of gas thickeners on gas diffusivity coefficient and gravity drainage mechanism is not
97 reported. Moreover, several experimental and theoretical studies are conveyed in the literature to
98 study the efficacy of a gas or solvent injection in NFRs under static conditions. Still, few pieces
99 of work have focused on the mass transfer mechanisms between matrix and fracture under
100 dynamic conditions. In addition, the gas diffusion coefficient is still a controversial challenge
101 during gas-based EOR for improving the gravity drainage mechanism in the gas invaded zone.
102 However, the gas thickener scenario can be one of the new approaches for improving the mass
103 transfer processes under flow conditions between fracture and matrix. Recently, several pieces of
104 research have been done to improve the oil recovery by polymer thickeners for gas-based EOR
105 in the conventional reservoir ([Gandomkar et al., 2021](#); [Gandomkar et al., 2020a](#); [Dai et al., 2018](#);
106 [Alhinai et al., 2017](#); [Lee et al., 2016](#); [Zhang et al., 2011](#)). The dissolution of polymer thickeners
107 in gases can cause a series of gas properties changes such as IFT reduction, gas diffusivity
108 coefficient improvement, and viscosity enhancement, which can promote the gravity drainage
109 mechanism in the gas invaded zone. Therefore, in this study, this new approach was considered
110 to investigate the effect of gas thickener on gravity drainage mechanism in the gas invaded zone.
111 Furthermore, polymer thickeners can improve gas characteristics in two ways: 1) by dissolving
112 heavy polymers (high molecular weight) in gases, and 2) by dissolving the small molecules as
113 direct thickeners. The substantial amount of co-solvent (toluene) in the case of heavy gas
114 thickeners, on the other hand, is troublesome and makes field application of this mixture
115 unfeasible. Furthermore, fluorine-based thickeners with high molecular weights are still the only
116 agents shown to dissolve in CO_2 without the need of a co-solvent. Therefore, except for the
117 fluorine based thickeners, it is adequate to utilise low molecular weight thickeners (without

118 adding co-solvents), as these are more economically agents during gas-based enhanced oil
119 recovery (Enick et al., 2012 and 2018; Dhuwe et al., 2016). Though, according to previous
120 studies, it was indicated that several gas thickeners such as poly(ethylene
121 oxide) (PEO), poly(vinyl alcohol) (PVOH), poly(styrene) (PS), poly(phenylene oxide) (PPO),
122 poly(acrylic acid) (PAA), poly(hydroxy alkanooates) (PHAA), poly(vinyl acetate) (PVAC), and
123 poly(isobutylene) (PIB), are seriously CO₂-phobic agents. On the contrary, poly(dimethyl
124 siloxane) (PDMS) and poly(fluoroacrylate) (PFA) are significantly CO₂-philic candidates (Xu et
125 al., 2001; Kikic et al., 2009; Mohamed et al., 2011; Enick et al., 2012 and 2018; Zaber et al.,
126 2020; Mao et al., 2013; Talebian et al., 2014; Gandomkar et al., 2021). Therefore, in this study,
127 PFA was used as a CO₂-philic gas thickener to improve the gravity drainage mechanism during
128 gas injection in the gas invaded zone. PFA is an amorphous, viscous, clear homopolymer that
129 dissolves in CO₂ at temperatures and pressures that are appropriate for CO₂-assisted oil recovery.
130 Furthermore, there is no extensive investigation of the mass transfer process in the literature for
131 CO₂-PFA thickener injection. Therefore, the goal of this paper is to recognize the governing oil-
132 recovery mechanisms during CO₂ thickener injection in the gas invaded zone such as CO₂
133 diffusion coefficient, IFT, and gravity drainage.

134

135 **Materials and methods**

136 **• Rock and fluid properties**

137 All laboratory experiments employed reservoir crude oil with an API of 32 from one of the
138 Middle Eastern oil fields. The oil filtration was performed to separate particles and impurities
139 from the oil to reduce any experimental complications. In addition, original formation water
140 (172000 ppm) was considered for gravity drainage tests. The properties of both formation water

141 and reservoir crude oil are reported in [Table 1](#). Also, the carbonate reservoir rock was used for
142 gravity drainage and CO₂ diffusion coefficients tests. The chemical composition of carbonate
143 rock was determined using the XRD (X-Ray Diffraction) technique. According to the
144 observations, the crushed material contains roughly 80 % calcite (CaCO₃), 11 % dolomite
145 (CaMg(CO₃)₂), 5 % anhydrate (CaSO₄) and 4% clay. The corresponding error values for the
146 XRD results were lower than ± 0.5 %. In addition, during the gravity drainage and gas diffusion
147 coefficient processes, the connate water saturation and the wettability of the core is set to those
148 of their reservoir conditions.

149 **Table 1**

- 150 **• Cloud point measurement**

151 In this study, the poly(fluoroacrylate) gas thickener was synthesized by our team following the
152 synthesis that has been previously described in detail elsewhere ([Zaberi et al., 2020](#)); the average
153 molecular weight of the polymer was 550000 g/mol. PFA is a CO₂-philic agent which is
154 monomer based and has six fluorinated carbons (not eight), thus eradicating the environmental
155 concerns that are associated with possible degradation products. The gas thickeners solubility in
156 CO₂ was measured by the HPHT visual cell described in our previous works ([Gandomkar et al.,
157 2020a, 2020b; Azizkhani and Gandomkar, 2019](#)). However, a specific amount of PFA must be
158 weighed and injected into the window cell first. The sample was then given a certain amount of
159 carbon dioxide to achieve the appropriate composition. A magnetic stirrer was utilised to create a
160 revolving magnetic field from a pressurised mixture with a constant overall composition (2000
161 rpm). It was repeated until the window cell produced transparent, single-phase solutions at a
162 suitable temperature and pressure. Finally, all samples were subjected to pressure reductions at
163 intervals of 20 psi. The equilibrium condition took roughly two hours to identify any visual

164 changes, and the poor solubility thickeners may take more time. Generally, the cloud point
165 pressures of gas/thickeners in the fog form were determined in the bulk sample by visual
166 monitoring. The measurements were taken at least three times, with a ± 5 psi repeatability. This
167 process was implemented for different PFA concentrations such as 5000, 10000, 20000, and
168 30000 ppm. Next, the mixtures were used for all experiments, such as IFT measurements, CO₂
169 diffusion coefficient calculations, and gravity drainage tests.

170

171 • **IFT measurement**

172 The interfacial tension of oil/CO₂ and oil/CO₂-PFA were measured using the HPHT IFT 700
173 equipment. The pendant drop technique is state of the art and precise method for determining the
174 IFT. During IFT measurements, a drop of oil is formed from the capillary needle's tip, which
175 CO₂ or CO₂-PFA bounds at the reservoir conditions ($P_{res}=3000$ psi and $T_{res} = 100$ °C).
176 Furthermore, the IFT error was computed using the standard deviation of four repeat
177 measurements of each mixture and was around 0.1 ([Gandomkar et al., 2020b](#); [Azizkhani and](#)
178 [Gandomkar, 2019](#)).

179 • **CO₂ diffusion coefficients measurement in the matrix-fracture system**

180 [Figure 1](#) shows a schematic design of the experimental setup for determining CO₂ diffusion
181 coefficients. Diffusion cell, gas/oil supply system, HPLC pump, data gathering system, and
182 temperature maintenance system were the primary components. The fluids and porous media
183 were held in a diffusion cell with an ID of 5 cm and a depth of 10 cm. The diffusion cell was
184 built to withstand pressures of up to 8000 psi and temperatures of up to 150°C. In this study, the
185 gas thickener or CO₂ diffusivity coefficients in bulk oil and porous media are measured in the

186 same diffusion cell. Therefore, for the diffusivity of the gas in the oil, the diffusion cell is filled
187 with gas thickeners (or CO₂) and oil with a suitable contact interface between the oil and gas
188 thickeners. This scenario was created to test diffusivity without the use of porous media. In
189 addition, a core that was initially saturated was inserted into the diffusion cell in the instance of
190 gas diffusivity in porous media. As a result, the annulus space provided a larger region for gas
191 diffusion into the core. It simulating the situation in which CO₂/gas thickeners are injected into
192 gas invaded zone. In the case of the bulk oil phase, all the containers were cleaned, and then all
193 cylinders were vacuumed for two hrs. After that, a required volume of oil was pumped into the
194 diffusion cell. Gas thickeners or CO₂ were transferred to a cylinder, and then the HPLC pump
195 pressurised it to the desired pressure. The system was maintained at a desirable temperature for 2
196 hrs. At the beginning of the diffusion test, the pressurised gas thickeners or CO₂ was transferred
197 to the diffusion cell. The pressure was logged by the pressure transducer connected to data
198 acquisition in the cell. When the diffusion process achieved a steady-state condition, it came to
199 an end. Furthermore, the approach for the gas diffusion experiment on porous media was
200 identical to scenario 1. Rather than injecting the oil straight into the cell, the diffusion cell was
201 filled with a core saturated with formation fluids. The core saturation was carried out in a
202 separate coreflooding setup, described in our previous works in detail ([Gandomkar and](#)
203 [Rahimpour 2015, 2017](#)). The core is prepared during the gas diffusion coefficient process based
204 on the idea that the core's saturation state (connate water saturation) and wettability (aging
205 process) are restored to their original state. The core sample C5 from [Table 2](#) was used for the
206 gas diffusion coefficient process. The gas diffusion coefficient tests were repeated for pure CO₂
207 and CO₂/PFA in different PFA concentrations such as 5000, 10000, 20000, and 30000 ppm. The
208 pressure decay method is widely used to measure the CO₂/gas thickeners diffusion coefficient

209 and thus was applied in this work (Gao et al., 2019; Li et al., 2006). The CO₂ or gas thickener
 210 diffusion coefficient will be estimated by mathematical model based on the measured
 211 instantaneous pressure data. The mathematical description of CO₂ or gas thickener diffusion in
 212 the porous medium could be defined as follows based on the Fick's law (Li et al., 2016; 2009):

$$213 \quad \frac{\partial C(r,t)}{\partial t} = D_{eff} \frac{\partial^2 C(r,t)}{\partial r^2} \quad (1)$$

214 The boundary and initial conditions are:

$$215 \quad C(r,t) = 0 \text{ at } 0 < r < r_0 \quad (2)$$

$$216 \quad C(r,t) = 0 \text{ at } t \geq 0 \text{ and } r = r_0 \quad (3)$$

217 Where C(r,t) are the gas thickener or CO₂ concentration during the diffusion process, mol.m³; t is
 218 the diffusing time, s; D_{eff} is the effective diffusion coefficient, m².s⁻¹; r₀ is the core radii, m; and r
 219 is the CO₂ diffusion radius, 0 < r < r₀, m. The analytical solution to diffusion equation is:

$$220 \quad C = C_0 \left[1 - \frac{2}{r_0} \sum_{n=1}^{\infty} \frac{J_0(r a_n) \exp(-D_{eff} a_n^2 t)}{a_n J_1(r_0 a_n)} \right] \quad (4)$$

221 Changing equation (4) in the form of mass and integrating it with r, and then replacing the real
 222 gas equation of state ($\Delta PV = Z \Delta n RT$) into equation 4, could be presented in the form of the
 223 instant pressure change among the square root of time:

$$224 \quad \Delta P = \frac{4M_{\infty} Z R T \sqrt{D_{eff}}}{r_0 V \sqrt{\pi}} \sqrt{t} = k \sqrt{t} \quad (5)$$

225
 226 Where K can be calculated through the simple linear regression, the gas thickener or CO₂
 227 diffusion coefficient is determined by equation (6) (Chai et al., 2019; Li et al., 2006):

$$228 \quad D_{eff} = \frac{\pi}{16} \left(\frac{r_0 k V}{M_{\infty} Z R T} \right)^2 \quad (6)$$

229 Where:

- 230 ❖ r_0 : porous media's radii, m;
- 231 ❖ k : the gradient of the pressure change v.s the square root of time;
- 232 ❖ V : gas thickener or CO₂ volume in the annulus area between the core and the cell, m³;
- 233 ❖ M_∞ : CO₂ or gas thickener dissolved in the porous medium, mole;
- 234 ❖ Z : Gas thickener or CO₂ deviation factor, dimensionless;
- 235 ❖ R : gas constant,
- 236 ❖ T : temperature in Kelvin.

237
 238 Also, in the bulk oil phase, the molar flux of gas thickener or CO₂ diffusing into an oil column
 239 can be presented based on Fick's law (Hoteit et al., 2009; Zhang et al., 2000). The procedure is
 240 similar to the mathematical model described for porous media. Therefore, based on the Fick's
 241 law, the relationship between the pressure and time is:

$$242 \quad P(t) = P_{eq} + a_1 \exp(-b_1 t) + a_2 \exp(-b_2 t) \quad (7)$$

243 Where all the constants a_1 , b_1 , a_2 , b_2 and P_{eq} can be calculated through the non-linear regression
 244 of the experimental data. After that, the gas thickener or CO₂ diffusion coefficient in the bulk oil
 245 phase can be calculated as follows:

$$246 \quad D_{AB} = \frac{4b_1 H^2}{\pi^2} \quad (8)$$

247 where the liquid height in the cell is shown as H , m; and D_{AB} is the gas thickener or CO₂
 248 diffusion coefficient, m².s⁻¹.

249 **Figure 1**

- 250 • **Gravity drainage process in gas invaded zone**

251 The experimental setup, shown in [Figure 2](#), simulated the vertical gravity drainage mechanism in
 252 a matrix block-fractured system in gas invaded zone. The main parts of the experimental setup
 253 are a vertically-mounted core holder, BPR, oven, data acquisition, HPLC pump, transfer vessel,

254 and separator. The core holder has several pressure gauges to display the critical parameters
255 linked with oil recovery from the cores. A total length of 72.9 cm carbonate cores (4 carbonate
256 cores) with 4 inches (10.16 cm) in diameter was centered in the middle of a core holder with an
257 internal diameter of 11 cm; 0.84 cm larger than the carbonate core. The annular space of 0.42 cm
258 can simulate the experimental model's vertical fracture. It should be noted that oil recovery by
259 gravity drainage in gas invaded zone is a function of capillary continuity between matrix blocks.
260 Several authors have addressed this phenomenon reporting different results ([Firoozabadi et al.,](#)
261 [1990, 1994; Saidi, 1991](#)). Therefore, the horizontal fracture opening was considered 30 μm to
262 provide the capillary continuity between blocks in fractured reservoirs. Moreover, the carbonate
263 core samples were initially saturated by formation water ([Table 2](#)). Reservoir crude oil was then
264 injected to create connate water saturation. The core saturation procedure was made in a self-
265 governing coreflooding setup, described in our previous works in detail ([Gandomkar and](#)
266 [Rahimpour, 2015 and 2017; Gandomkar et al., 2013; Nematzadeh et al., 2012](#)). During gravity
267 drainage tests, the cores are also prepared with the intention of restoring the connate water
268 saturation and the wettability of the core (aging process) to their reservoir conditions
269 ([Zendehboudi et al., 2011](#)). After that, the saturated oil-wet cores were centered vertically in the
270 middle of a core holder. The system was obtained to the desired temperature, and then gas
271 thickener or CO_2 was inserted into the 0.42 cm wide annulus, to simulate the fractures between
272 the core samples and core holder. The system was pressurised by gas thickener or CO_2 injection
273 to the desired pressure, and finally, oil recovery versus time was documented. Other researchers
274 have already used the described model to simulate the oil recovery from gas invaded zone
275 experimentally ([Schechter et al., 1996; Pooladi-Darvish et al., 2000; Babadagli et al., 2003](#)). It
276 should be highlighted that oil recovery by gravity drainage in gas invaded zone significantly

277 depends on capillary and gravity forces. Therefore, to experimentally simulate the gravity
278 drainage process in the gas invaded zone, gas thickener or CO₂ was injected from the top of the
279 column at a constant frontal advance rate. At the same time, the capillary number was controlled
280 to be lower than its critical value (less than 10⁻⁵) to establish the capillary force dominating the
281 flow process (Babadagli et al., 2003; Firoozabadi et al., 1990, 1994; Saidi 1991). Two different
282 scenarios, including CO₂/PFA and pure CO₂ injection, were considered to study the impact of
283 gas thickener on oil recovery in the gas invaded zone. In addition, the oil recovery factors were
284 measured based on the original oil in place.

285 **Figure 2**

286 **Table 2**

287 **Results and discussion**

288 This study investigates the effect of CO₂-philic polymeric thickener (PFA) on the gravity
289 drainage mechanism in the gas invaded zone during CO₂ injection through the synergy of the
290 interfacial mechanisms. First, the dissolution of PFA in CO₂ was conducted for different PFA
291 concentrations, 5000, 10000, 20000, and 30000 ppm, via cloud point pressure measurements.
292 After that, these new resolutions were used for all other tests. Therefore, the effect of PFA-
293 thickened carbon dioxide on IFT measurements was estimated at various temperatures. After
294 that, the CO₂ diffusion coefficients in porous and non-porous media were calculated during pure
295 CO₂ and PFA-CO₂ scenarios. Finally, the oil recovery was illustrated through a gravity drainage
296 mechanism during CO₂ thickener injection in the gas invaded zone under reservoir conditions
297 (i.e. T_{res} = 100 °C and P_{res} = 3000 psi). These results have been presented as follows.

298

299

300

301 • **The dissolution of gas thickener in CO₂**

302 The dissolution of PFA in CO₂ was examined by calculating cloud point pressures. The cloud
303 point appears as the pressure at which the single-phase solutions were achieved at favourable
304 temperature and pressure. After that, the resulting single-phase mixtures were used for all other
305 tests in the gas invaded zone. The cloud point pressures of CO₂/PFA solutions with four different
306 PFA concentrations, 5000, 10000, 20000, and 30000 ppm, for temperatures of 40, 70, and 100 °C
307 were reported in [Figure 3](#). The cloud point pressure measurements were 2250 to nearly 3100 psi.
308 These results show that increasing temperature and PFA concentration generally raise the cloud
309 point pressures. These were 2260, 2420, and 2680 psi at different temperatures of 40, 70, and
310 100 °C, respectively, for 5000 ppm PFA concentration. Also, these were 2680, 2810, 2950, and
311 3100 psi for different concentrations of 5000, 10000, 20000, and 30000 ppm, respectively, at 100
312 °C. The results highlighted that the lower temperatures provided higher thickeners solubility in
313 CO₂. Moreover, the high concentrations of PFA (30000 ppm) increased the cloud point pressures
314 to 3100 psi at 100 °C. In addition, [Figure 4](#) illustrates the effect of PFA concentrations on the
315 cloud point pressure at 40, 70, and 100 °C. From this result, it increases approximately linearly
316 with increasing in PFA concentrations. Also, it showed that the cloud point pressure was higher
317 than it by increasing the temperature at the same PFA concentration. It is known as the entropy
318 of mixing, and it has the ability to control this condition. The density of PFA is almost
319 unchanged with temperature, whilst the density of gas rises as temperature decreases. As a result,
320 the density difference widens and the entropy of mixing decreases, causing the temperature to
321 behave inversely during the thickened gas phase ([Azizkhani and Gandomkar 2019](#)). Enick et al.
322 ([2018](#)) showed that the PFA is remarkably soluble in CO₂, requiring only about 1450 psi to

323 dissolve 30000 ppm of PFA in CO₂ at 24°C. The main difference between their results and our
324 measurements is referred to as PFA molecular weight and temperature conditions. However, our
325 cloud point measurements have a good consistency with their results. PFA solubility in CO₂ was
326 determined, and single-phase solutions were employed for all subsequent studies in the gas
327 invaded zone. As a result, in the remaining trials in this work, the pressure was always kept
328 above the cloud point pressure to guarantee that the PFA did not come out of the solution
329 because of phase change.

330 **Figure 3**

331 **Figure 4**

332 **• The effect of gas thickener on IFT**

333 Karimaie et al. (2010) investigated the gravity drainage mechanism in fractured carbonate rock
334 during gas injection in low IFT. They reported that the gas-oil gravity drainage at low IFT is an
335 efficient oil recovery technique at secondary and tertiary injection in the gas invaded zone.
336 Therefore, in this study, the CO₂/PFA solutions were considered to examine the performance of
337 gas thickeners on gravity drainage by IFT reduction. The impact of PFA thickener on oil and
338 CO₂ interfacial tension was measured at reservoir pressure (3000 psi). Also, the cloud point
339 pressure for 30000 ppm PFA at 100 °C was higher than reservoir pressure (3100 psi). Therefore,
340 only at this point, the IFT was conducted at a pressure higher than reservoir pressure (3200 psi)
341 to guarantee that the PFA did not come out of the solution due to a phase change. The results
342 (Table 3) displays that the high molecular weight of PFA can meaningfully decrease the IFT. For
343 example, the IFT between the pure CO₂ and reservoir fluid was 56 dyne/cm, and it was lowered
344 to 24 dyne/cm for 20000 ppm CO₂/PFA scenario at reservoir conditions. Also, the IFTs were

345 increased by increasing temperature, but while PFA dissolved to the CO₂, it was increased lower
346 than that compared to pure CO₂ scenarios. Moreover, [Figure 5](#) illustrates the effect of PFA
347 concentrations on IFTs between CO₂/PFA and reservoir crude oil at the reservoir conditions. The
348 findings indicated that the IFTs were reduced by increasing PFA concentrations. It could be
349 considered as a rise in gas density in the attendance of thickeners ([Harrison et al., 1996](#)). [Figure 6](#)
350 shows the MMPs of crude oil and CO₂/thickener estimated using the vanishing interfacial
351 tension (VIT) approach using interfacial tension data ([Ghorbani et al. 2019 and 2020;](#)
352 [Gandomkar et al., 2020b](#)). The MMPs were 3510 and 3320 psi for pure CO₂ and 20000 ppm
353 PFA, respectively, at reservoir temperature. The results show that the CO₂ thickener could
354 meaningfully decrease the minimum miscibility pressure. Consequently, the MMP for pure
355 carbon dioxide and PFA-thickened CO₂ (20000 ppm) were more than the reservoir pressure
356 ($P_{res}=3000$ psi). So the immiscible injection will happen under reservoir conditions during the
357 gravity drainage and gas diffusion coefficient experiments.

358 **Table 3**

359 **Figure 5**

360 **Figure 6**

361 **• Gas diffusion in matrix block during CO₂/thickener injection**

362 The gas diffusion coefficient is a key parameter to control the mixing rate of the injected gas and
363 crude oil. By injecting gas into the gas invaded zone, gas associates with the oil through
364 diffusion, that result in change in different properties such as drop in IFT and viscosity, and oil
365 swelling, which can improve the oil recovery in the gas invaded zone. Thus, it is vital to explore
366 the gas diffusivity under real reservoir conditions during CO₂/PFA scenarios. [Table 4](#) shows the
367 gas diffusion coefficients for the CO₂–oil and CO₂/PFA–oil systems in porous media and bulk oil

368 phase scenarios. At reservoir conditions, the pure CO₂ diffusion coefficients in the bulk oil phase
369 and porous media were 11.8 and 0.24 * 10⁻⁹ m².s⁻¹, respectively. Results showed that the pure
370 CO₂ diffusion coefficient in the bulk oil phase was higher than that in porous media. In other
371 words, the CO₂ pressure drop in the bulk oil phase system was more significant than that in
372 porous media, demonstrating a higher volume of CO₂ was dissolved. The enlargement of the
373 contact area between oil and CO₂ in bulk oil phase system compared to porous media system
374 significantly improved the gas diffusivity coefficient. Therefore, mass transfer between fracture-
375 matrix decreases, resulting in low CO₂ reaching inside the porous media compared to the oil
376 phase system. In the porous media case, the mass transfer was decreased due to heterogeneity
377 compared to the bulk oil phase. CO₂ penetrates the thin oil film first and then disperses the oil in
378 the porous media. Furthermore, the pores of varied diameters are twisted and interlinked in
379 carbonate reservoirs due to the complex geological sedimentation processes. The path for a gas
380 molecule to diffuse via the pores is complex and tortuous. As a result, reaching the inside porous
381 media would take the longest, corresponding to the pressure decrease in the system (Zhang et al.,
382 2000; Chai et al., 2019). Based on Figure 7, the CO₂ diffusion coefficients were increased during
383 PFA/CO₂ injection in both scenarios. For example, these were 33.8 and 0.56 * 10⁻⁹ m².s⁻¹ for
384 20000 ppm polymer thickener in the bulk oil phase and porous media systems, respectively, at
385 reservoir conditions. Renner (1988) illustrated that the diffusion coefficient is extremely reliant
386 on both solvent and solute viscosity (oil and CO₂ respectively), that highlights the impact of
387 temperature, pressure and fluid composition. Therefore, it can be rationalised by bearing in mind
388 two distinctive phenomena from the aspect of the gas thickener enhanced oil recovery technique;
389 1) a reduction in oil viscosity due to CO₂ diffusion, and 2) increasing gas viscosity due to
390 polymer thickener dissolution in CO₂. Additionally, these two phenomena may improve the CO₂

391 diffusion coefficient during CO₂/PFA injection. During the gas-based enhanced oil recovery, the
392 porous media's properties are of great importance in analysing the gas diffusion process in a
393 fractured-matrix system. Therefore, the relationship of CO₂ diffusion between the bulk oil phase
394 and porous media can be described as the effective diffusion coefficient (Li et al., 2006; Hoteit et
395 al., 2009):

$$396 \quad D_{eff} = \frac{\varphi D_{bulk}}{\tau} \quad (9)$$

397 Where:

- 398 ❖ D_{eff} : gas diffusion coefficient in porous media;
- 399 ❖ D_{bulk} : gas diffusion coefficient in the bulk oil phase;
- 400 ❖ φ : porosity;
- 401 ❖ τ : tortuosity of the porous media;

402 **Table 4**

403 **Figure 7**

404 • **Enhanced oil recovery in gas invaded zone during CO₂/thickener injection**

405 Gravity drainage is one of the most critical mechanisms in gas invaded zone, and it plays a
406 significant impact on oil recovery during gas-based methods. Moreover, the synergy of the
407 aforementioned mechanisms on oil recovery was investigated by two different sets of gravity
408 drainage scenarios: pure CO₂ and PFA/CO₂ (20000 ppm) injection. Figure 8 illustrates the
409 ultimate oil recovery factor based on the gravity drainage process in the gas-vented zone at
410 reservoir conditions. The pure CO₂ and PFA/CO₂ scenarios indicated that about 36 and 52
411 percent oil recovery factor was produced in the gas invaded zone during the gravity drainage
412 process. Therefore, an incremental oil recovery of 16 percent was achieved during PFA/CO₂
413 compared to pure CO₂ injection in the gas invaded zone. The lower oil recovery achieved in the

414 pure CO₂ scenario is due to a high capillary hold-up zone. The density difference and low
415 interfacial tension between the phases also contribute to the substantial oil recovery efficiency
416 recorded during the gravity drainage test. In this study, the gas and oil density to increase by
417 dissolution of PFA in CO₂ and also more gas in solution, respectively which caused a change in
418 density difference and consequently improving oil recovery. The principal force countered by the
419 matrix capillary pressure is the density difference between pure CO₂ or CO₂/PFA in the fracture
420 and the oil in the matrix. Furthermore, the interfacial tension between oil and CO₂/PFA lowers,
421 resulting in a decrease in capillary hold-up. The remaining oil saturation above the new hold up
422 zone will also diminish in this instance, especially if the interfacial tension goes below one
423 dyne/cm (Karimaie et al., 2010). Additionally, it should be highlighted that in cases where high
424 CO₂ solubility in the oil phase increases the oil density, the higher density leads to unavoidable
425 natural convection and, consequently, higher oil recovery. Accordingly, the high CO₂ mass
426 transfer and IFT reduction during the gas thickener injection compared to pure CO₂ scenario can
427 improve the oil recovery in gas invaded zone. This is consistent with the findings of other
428 researchers (Pedrera et al., 2002; Shahidzadeh et al., 2003).

429 **Figure 8**

430 **Conclusion**

- 431 • Lower temperatures provided higher PFA solubility in CO₂. Also, the high concentration
432 of PFA (30000 ppm) increased the cloud point pressure to 3100 psi at 100 °C.
- 433 • The high molecular weight of PFA thickener could enhance the oil recovery in gas
434 invaded zone due to a decrease in the IFTs.
- 435 • The CO₂ thickener could significantly reduce the minimum miscibility pressure and may
436 be improved oil recovery during the gravity drainage process.

- 437 • The CO₂ pressure drop in the bulk oil phase system was more significant than that in
438 porous media, indicating a higher amount of CO₂ diffusion coefficient.
- 439 • The mass transfer was decreased due to heterogeneity compared to the bulk oil phase
440 during the CO₂ diffusion tests.
- 441 • The CO₂ diffusion coefficients were increased during PFA/CO₂ injection in both
442 scenarios at reservoir conditions.
- 443 • An incremental oil recovery of 16 percent was achieved during PFA/CO₂ compared to
444 pure CO₂ injection in the gas invaded zone.

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460 **Nomenclature**

461	API	American Petroleum Institute
462	EOR	Enhanced Oil Recovery
463	HPHT	High Pressure-High Temperature
464	HPLC	High Pressure Liquid Chromatography
465	IFT	Interfacial Tension
466	MMP	Minimum Miscibility Pressure
467	PAA	poly(acrylic acid)
468	PDMS	poly(dimethyl siloxane)
469	PEO	poly(ethylene oxide)
470	PFA	poly(fluoroacrylate)
471	PHAA	poly(hydroxy alkanates)
472	PIB	poly(isobutylene)
473	PPO	poly(phenylene oxide)
474	P_{res}	Reservoir Pressure
475	PS	poly(styrene)
476	PVAC	poly(vinyl acetate)
477	PVOH	poly(vinyl alcohol)
478	rpm	Revolutions Per Minute
479	T_{res}	Reservoir Temperature
480	VIT	Vanishing Interfacial Tension
481	XRD	X-Ray Diffraction

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657 **List of Tables**

658 **Table 1:** The properties of both formation water and reservoir crude oil

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661 temperatures

662 **Table 4:** The gas diffusion coefficients measurements in both bulk oil phase and porous media
663 systems in different PFA concentrations at reservoir conditions

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Table 1

Formation water		Crude oil	
Ions	Concentrations (ppm)	Hydrocarbon Type	mole percent
Na ⁺	57441	N.Paraffins	36.4
Mg ²⁺	1783	Iso.Paraffins	21.5
Ca ²⁺	9704	Naphthenes	24.7
Cl ⁻	103021	Aromatics	15.2
HCO ₃ ⁻	28	Saturates C ₁₅ ⁺	1.2
SO ₄ ²⁻	6	Aromatics C ₁₅ ⁺	1.0
K ⁺	13	Total sum	100.0
Br ⁻	4	S.G (60°F), ASTM D40452	0.8
TDS (ppm)	172000	Molecular weight, g/mol, IP-86	106.3

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Table 2

Carbonate cores	Length (cm)	D (in)	PV (cc)	Porosity (%)	Permeability (md)	S_{wc} (%)
C1	15.7	4	145.1	11.4	5.8	27.5
C2	17.1	4	188.5	13.6	6.4	28.3
C3	18.7	4	159.2	10.5	6.8	29.0
C4	21.4	4	196.1	11.3	7.5	28.2
C5*	6.3	1.5	7.8	10.6	6.2	27.6

*This core was used to CO₂ diffusion coefficients tests in porous media

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Table 3

Injection Gas	PFA (ppm)	IFT (dyn/cm)		
		40 °C	70 °C	100 °C
CO ₂	0	30.0	42.0	56.0
CO ₂ /PFA	5000	23.0	31.0	41.0
	10000	14.0	26.0	33.0
	20000	05.0	12.0	24.0
	30000	00.5	06.0	15.0*

747 *IFT was conducted at pressure higher than reservoir pressure (3200 psi) to ensure that the PFA did not come out of
748 the solution due to a phase change. Because of the cloud point pressure of these conditions (30000 ppm and 100 °C)
749 was higher than reservoir pressure (3100 psi).

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Table 4

Scenarios	Gas diffusion coefficients ($10^{-9} \text{ m}^2.\text{s}^{-1}$)	
	Bulk oil phase	Porous media
Pure CO ₂	11.80	0.24
CO ₂ /PFA (5000 ppm)	21.10	0.41
CO ₂ /PFA (10000 ppm)	28.40	0.50
CO ₂ /PFA (20000 ppm)	33.80	0.56
CO ₂ /PFA (30000 ppm)	36.70	0.61

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813 **Figure 1:** A schematic diagram of the experimental setup for measuring the CO₂ diffusion
814 coefficients in both bulk oil phase and porous media systems at reservoir conditions

815 **Figure 2:** The experimental setup for gravity drainage mechanism in matrix block-fractured
816 system in gas invaded zone

817 **Figure 3:** The cloud point pressures of CO₂/PFA solutions with four different PFA
818 concentrations, 1: 5000 ppm; 2: 10000 ppm; 3: 20000 ppm; 4: 30000 ppm

819 **Figure 4:** The effect of PFA concentrations on the cloud point pressure at 40, 70, and 100 °C

820 **Figure 5:** The effect of PFA concentrations on IFTs between CO₂/PFA and reservoir crude oil at
821 reservoir conditions

822 **Figure 6:** The interfacial tension data for calculating the MMPs of crude oil and CO₂/thickener
823 by vanishing interfacial tension (VIT) technique

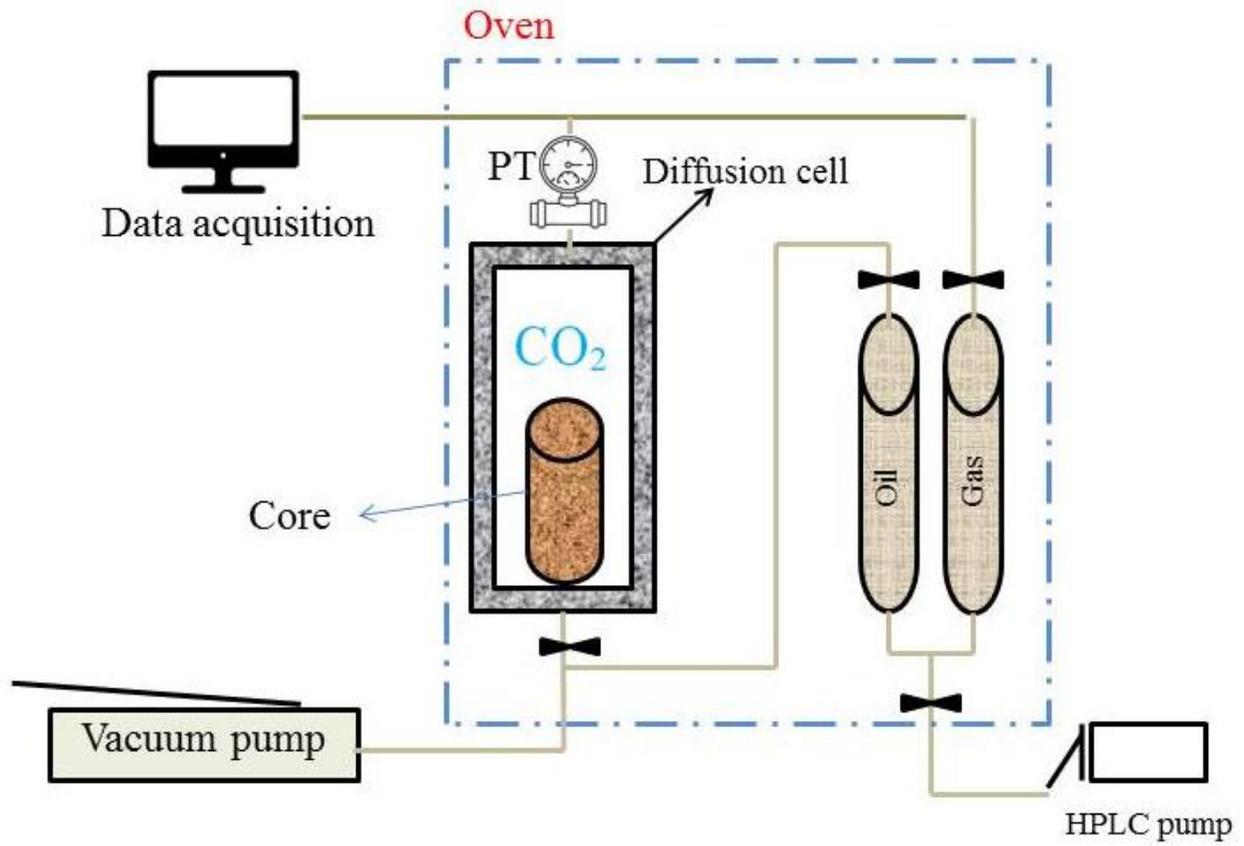
824 **Figure 7:** The CO₂ diffusion coefficients during PFA/CO₂ injection in both porous media and
825 bulk oil phase scenarios, 1: pure CO₂; 2: CO₂/PFA (5000 ppm); 3: CO₂/PFA (10000 ppm); 4:
826 CO₂/PFA (20000 ppm); 5: CO₂/PFA (30000 ppm)

827 **Figure 8:** The ultimate oil recovery factor during gravity drainage process in gas invaded zone
828 for pure CO₂ and PFA/CO₂ (20000 ppm) scenarios

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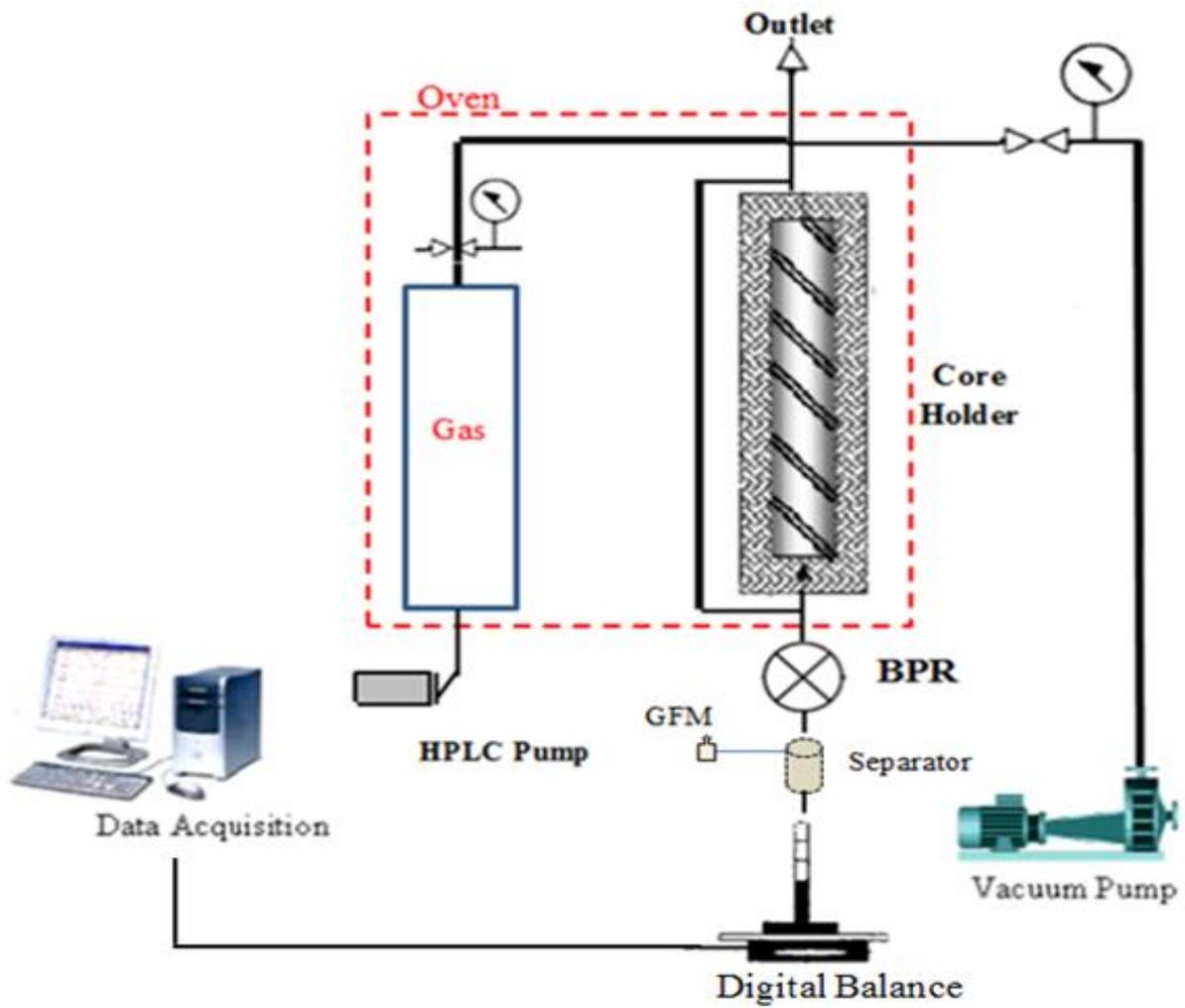
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Figure 1

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Figure 2

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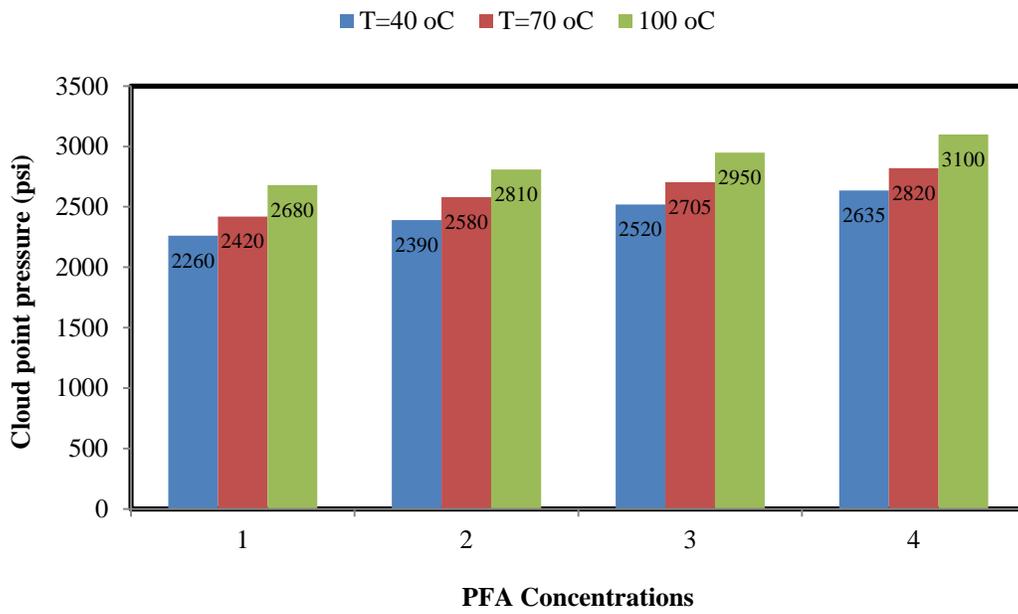
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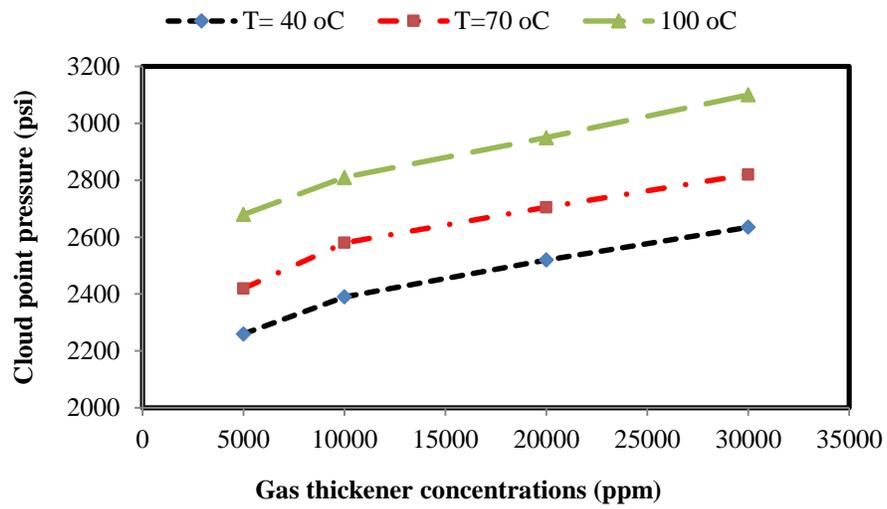
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Figure 4

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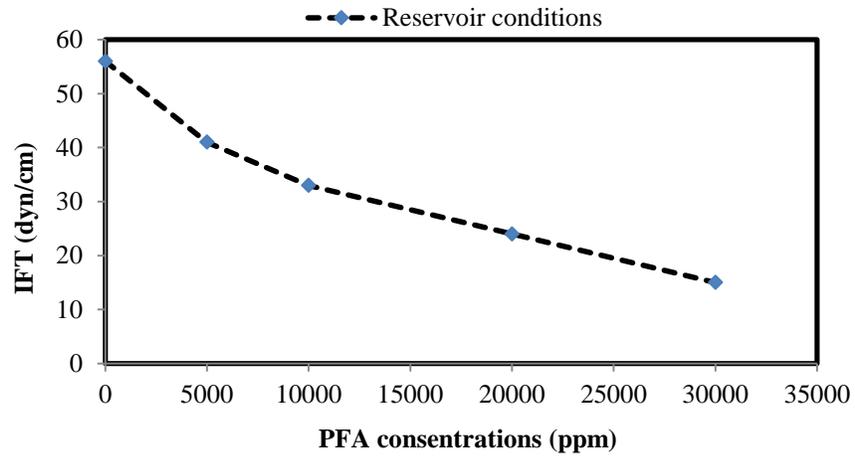
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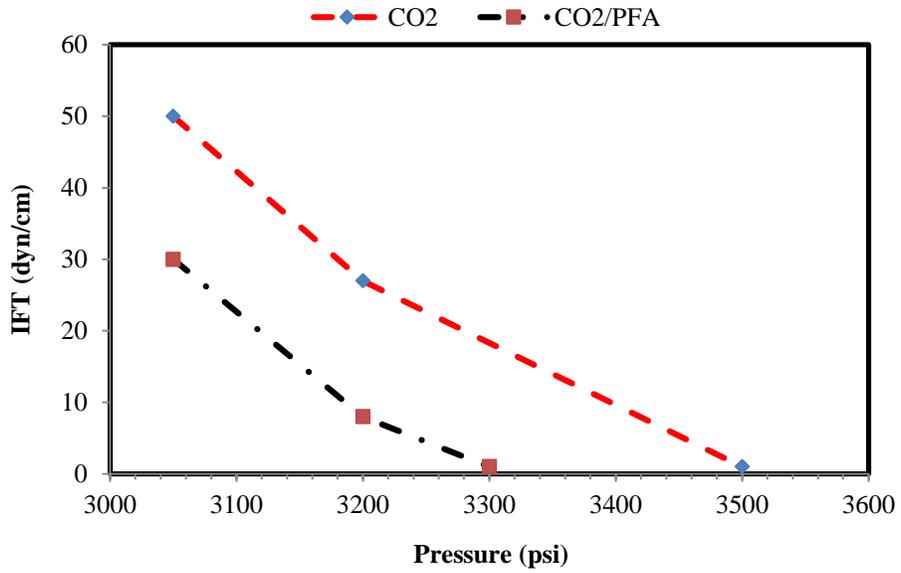
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Figure 5

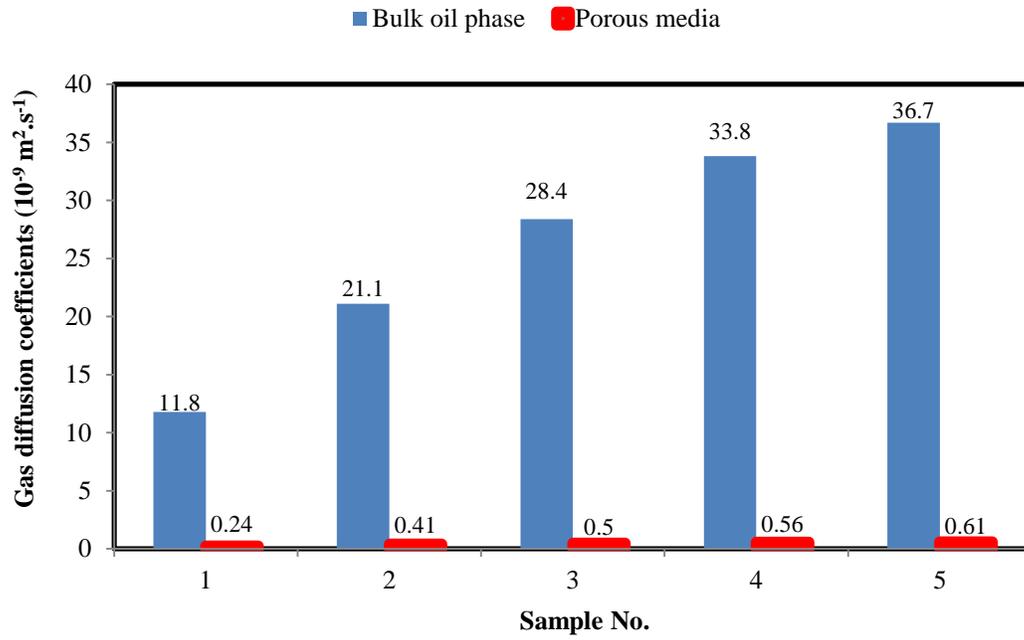
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Figure 6

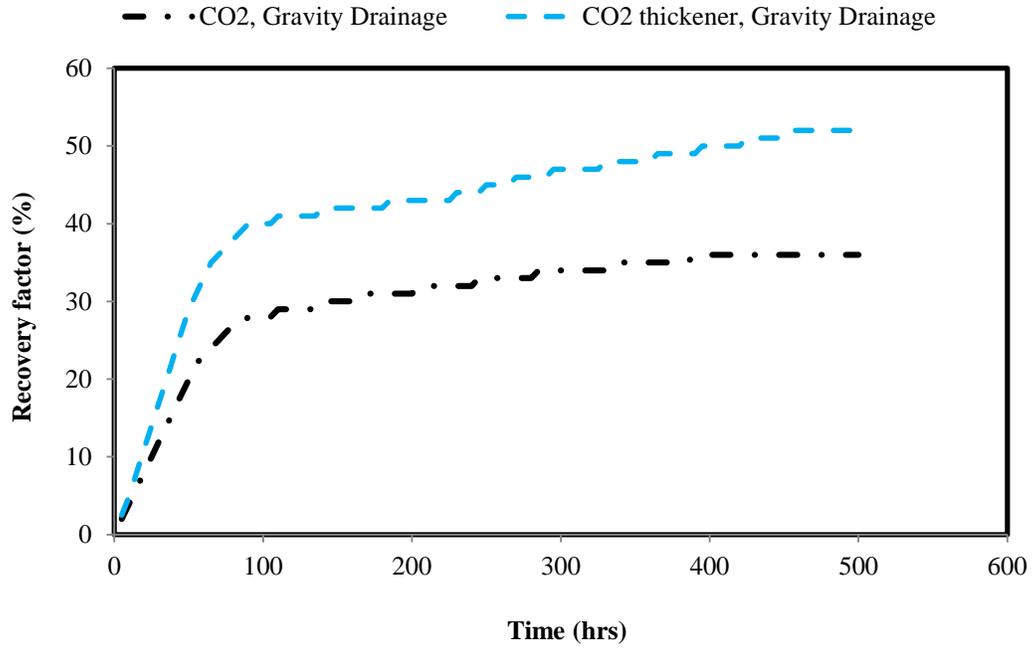
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Figure 7

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Figure 8

Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

Credit author statement

Asghar Gandomkar: Conceptualization, Methodology, Validation, Investigation, Resources, Writing - Original Draft, Writing - Review & Editing, Visualization, Supervision, Project administration. **Hamidreza Nasriani:** Conceptualization, Methodology, Validation, Investigation, Resources, Writing - Original Draft, Writing - Review & Editing, Visualization. **Robert M. Enick:** Conceptualization, Methodology, Investigation, Writing - Original Draft, Writing - Review & Editing. **Farshid Torabi:** Conceptualization, Methodology, Validation, Investigation, Resources, Writing - Original Draft, Writing - Review & Editing, Visualization.